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Studies toward the construction of substituted piperidine-2-ones and pyridine-2-ones from Baylis–Hillman adducts: discovery of a facile synthesis of 5-methyl-4-oxo-6-aryl-3-azabicyclo[3.1.0]hexane-1-carboxylates[☆]

V. Singh,^a G. P. Yadav,^b P. R. Maulik^{b,†} and S. Batra^{a,*}

^aMedicinal Chemistry Division, Central Drug Research Institute, PO Box 173, Lucknow 226001, India ^bMolecular and Structural Biology Division, Central Drug Research Institute, PO Box 173, Lucknow 226001, India

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Abstract—Studies toward the construction of functionalized piperidone derivatives from derivatives of Baylis–Hillman adducts are described. Interestingly the 6-oxo-4-aryl-piperidine-3-carboxylates generated during the study serve as precursor for the facile synthesis of 4-oxo-6-aryl-3-aza-bicyclo[3.1.0]hexane-1-carboxylates. © 2006 Elsevier Ltd. All rights reserved.

1. Introduction

The multifunctional nature of the backbone of Baylis-Hillman adducts provides an excellent opportunity to generate a variety of heterocycles employing simple synthetic manipulations. Notably, the last decade has witnessed an extraordinary growth in the number of reports describing different approaches to achieve the syntheses of an array of cyclic compounds using Baylis-Hillman chemistry.¹ Recently, we too have reported the synthesis of a variety of heterocycles in solution and on solid phase utilizing Baylis-Hillman chemistry.² In our continuing efforts aimed at this objective, we describe herein the results of our studies toward the synthesis of substituted piperidine-2-ones and pyridine-2ones from the nucleophilic substitution reaction products afforded by the reaction between the acetyl derivatives of Baylis-Hillman adducts and ethyl cyanoacetate. We have discovered that the substituted piperidine-2-one generated during the endeavor may serve as precursor for the efficient synthesis of 5-methyl-4-oxo-6-aryl-3-aza-bicyclo[3.1.0]hexane-1-carboxylates.

The piperidine ring system is a structural component of numerous naturally occurring alkaloids, biologically active synthetic molecules, and organic chemicals. The syntheses of piperidones and piperidines have been exhaustively reviewed recently by Sabol and co-workers.³ We envisaged the synthesis of piperidine-2-ones from the nucleophilic addition products afforded by reactions between acetates of the Baylis–Hillman adducts and ethyl cyanoacetate by reduction of the cyano group and subsequent intramolecular cyclization between the amino group and the ester moiety. On the other hand, the conversion of the cyano group to an amide in the same substrate should lead to 3-methylene-piperidine-2,6-diones (Fig. 1). The substituted piperidine-2,6-diones are structural component of several natural products and biologically active molecules.⁴



Figure 1. Strategy for the synthesis of substituted piperidine-2-ones and piperidine-2,6-diones.

2. Results and discussion

The preparation of the starting materials in our synthetic sequence (Scheme 1, acetates 3a-g) was accomplished by

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Keywords: Baylis–Hillman; Piperidine-2-one; DBU; 4-Oxo-3-aza-bicy-clo[3.1.0]hexane; 3-Methylene-piperidine-2,6-dione.

^{*} Corresponding author. Tel.: +91 522 2262411–18x4368; fax: +91 522 2623405/938; e-mail: batra_san@yahoo.co.uk

[†] For queries related to X-ray write to maulik_prakas@yahoo.com

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Scheme 1. Reagents and conditions: (i) CH_2 =CHCO₂Me, DABCO, rt, 2–5 days; (ii) AcCl, pyridine, CH_2Cl_2 , rt, 4–6 h; (iii) CNCH₂CO₂Et, DABCO, THF:H₂O, rt, 2 h; (iv) Raney-Ni, H₂, 40 psi, rt, 3 h; (v) DDQ, dioxane, reflux, 24 h; (vi) POCl₃, neat, reflux, 24 h; (vii) PCl₅, POCl₃, reflux, 2 h; (viii) DBU, CH₃CN, reflux, 14 h; (ix) FeCl₃·6H₂O, propionic acid, reflux, 2 h; (x) TFA:H₂SO₄, neat, rt, 3 h; (x) NaH, toluene, rt, 30 min.

acetylating the Baylis-Hillman adducts 2a-g, which in turn were obtained from substituted benzaldehydes (1a-g) following the literature procedure.⁵ The nucleophilic substitution of the acetates 3a-g with ethyl cyanoacetate in the presence of DABCO in a THF:water system following a standard procedure yielded the substituted 1,5-dipentanoate derivatives 4a-g as diastereoisomeric mixture in excellent yields.^{2a} As would be expected, the reduction of these compounds in the presence of Raney-nickel under hydrogenation conditions yielded 5-methyl-6-oxo-4-aryl-piperidine-3-carboxylic acid ethyl esters 5a-g in 54-67% yields. These compounds were obtained as mixtures of diastereoisomers. In principle, oxidation of these piperidinones with a suitable reagent should furnish pyridine-2-one derivatives 6. Accordingly, the aromatization of these compounds was investigated in the presence of DDQ.⁶ However, in our hands, the desired oxidation did not take place and only the starting material was recovered.

At this point, we envisaged that if compound **5** was converted to its chloro-derivative **7** with POCl₃ it would easily afford the desired pyridinone through oxidation.⁷ Unfortunately, chlorination in the presence of POCl₃ under several conditions failed to afford the chloro-derivative **7**. Subsequently, the chlorination of compound **5** was attempted with a mixture of PCl₅ and POCl₃. Interestingly, this reaction

yielded a less polar product 8, the structure of which was established on the basis of spectroscopic analysis. The general nature of halogenation was confirmed by the synthesis of compounds 8a-g. In principle, the tertiary nature of the chloro-group in compound 8 should make it an appropriate substrate for dehydrohalogenation followed by oxidation. In order to achieve the envisaged product, several reactions were attempted. It was gratifying to note that compounds **8a-g** undergo reaction in the presence of DBU to furnish products in good vields. On the basis of spectroscopic evidence the structure of these compounds was identified as 5-methyl-4-oxo-6-aryl-3-aza-bicyclo[3.1.0]hexane-1-carboxylates 9a-g. The structure of these products was ascertained unambiguously via X-ray analysis of a representative compound **9e** (Fig. 2).⁸ The formation of these products could be explained on the basis of the fact that the hydrogen atom on the carbon bearing the alkoxycarbonyl group being more acidic participates in the elimination of the chloride ion.

In a different strategy it was envisaged that the conversion of the cyano group to an amide would lead to an intermediate, which should undergo intramolecular cyclization resulting in 3-methylene-piperidine-2,6-dione. Recently Wang et al. have described an interesting FeCl₃-mediated highly efficient synthesis of 1,2-dihydro-2-oxo-3-pyridine-carboxylate



Figure 2. ORTEP diagram showing the crystal structure of 9e with atomic numbering scheme for non-H atoms only at 30% probability level.

starting from enones and ethyl cyanoacetate.9 Following their strategy, treatment of compounds 4b,e,g with 3 equiv of FeCl₃·6H₂O in propionic acid at reflux for 2 h afforded 3-methylene-4-substituted phenyl-piperidine-2,6-diones 10b,e,g in 60-65% yields. However, in contrast to their results, we observed that the carboxylate group was lost during the reaction and dehydrogenation did not occurred. In order to investigate the scope of our substrates for the generation of the carboxylate group containing piperidine-2,6-dione derivative, 4a,e,f were hydrolyzed in the presence of TFA:H₂SO₄ mixture to afford the products **11a.e.f** in good vields. Subsequent treatment of the amides 11a.e.f with NaH at room temperature furnished the desired 5-methylene-2,6-dioxo-4-phenyl-piperidine-3-carboxylic acid ethyl esters 12a.e.f in good yields. Interestingly, the formation of compounds 12a,e,f was highly stereoselective in favor of the trans-isomer. The stereochemistry was assigned on the basis of selective 1D NOESY experiment with compound 12a. During the progress of this work, Kim et al. reported the synthesis of 3,5-dimethylene-4-phenyl-piperidine-2,6-dione and mono-alkylidene glutarimide by hydrolysis of the cyano group with sulfuric acid in methanol followed by cyclization in the presence of sodium bicarbonate.¹⁰ However, in our hand, the sodium bicarbonate-mediated cyclization of substrates **11a.e.f** led to a complex mixture of product from which the corresponding 3-methylene-piperidine-2,6-diones 12a,e,f were isolated in only low yields.

3. Conclusions

In summary, we have demonstrated a convenient and general process for the synthesis of 5-methyl-4-oxo-6-aryl-3-aza-bi-cyclo[3.1.0]hexane-1-carboxylates via products afforded by the nucleophilic addition reaction of ethyl cyanoacetate with acetyl derivatives of the Baylis–Hillman adducts. Additionally these products have been readily hydrolyzed and subsequently cyclized to yield new substituted piperidine-2,6-diones in good yields.

4. Experimental

4.1. General

Melting points were uncorrected and determined in capillary tubes on a hot stage apparatus containing silicon oil. IR spectra were recorded using a Perkin–Elmer RX I FTIR spectrophotometer. ¹H and ¹³C NMR spectra were recorded on either 300 or 200 MHz FT spectrometer, using TMS as an internal standard (chemical shifts in δ values, *J* in hertz). The FABMS were recorded on a JEOL/SX-102 spectrometer and ESMS were recorded in a Micromass LCMS system. Elemental analyses as performed on a Carlo Erba 1108 microanalyzer or Elementar's Vario EL *III* microanalyzer. All compounds were obtained in the powder form unless otherwise stated.

4.2. General procedure for the synthesis of compounds **4a–c**, as exemplified for compound **4a**

To a stirred solution of compound 3a (2.0 g, 8.5 mmol) in THF:H₂O (20 mL, 50:50, v/v) was added DABCO (1.5 g,

12.8 mmol) at room temperature and the reaction was allowed to continue for 20 min. Thereafter, ethyl cyanoacetate (1.1 mL, 10.3 mmol) was added to the reaction mixture and the reaction was allowed to proceed at room temperature for 2 h. Then THF was removed from the reaction mixture via rotary evaporation and the residue was diluted with water (100 mL) and extracted with EtOAc (3×40 mL). The organic layers were pooled, washed with brine (50 mL), dried (Na_2SO_4), and evaporated in vacuo to yield a residue, which was purified by silica gel chromatography employing hexane:EtOAc (80:20, v/v) to afford 2.0 g (82%) of product **4a** as colorless oil.

4.2.1. 2-Cyano-4-methylene-3-phenyl-pentanedioic acid-1-ethyl ester-5-methyl ester (4a). ν_{max} (neat) 1747 $(CO_2Me \text{ and } CO_2Et)$, 2255 $(CN) \text{ cm}^{-1}$; ¹H NMR $(CDCl_3)$, 200 MHz) $\delta = 1.15$ (t, 3H, J = 7.1 Hz, CH_3CH_2), 1.33 (t, 3H, J=7.1 Hz, CH₃CH₂), 3.71 (s, 3H, CO₂CH₃), 3.76 (s, 3H, CO₂CH₃), 4.10–4.30 (m, 6H, $2 \times CH_2$ CH₃ and 2×CHAr), 4.45 (d, 1H, J=8.2 Hz, CHCN), 4.64 (d, 1H, J=8.2 Hz, CHCN), 5.75 (d, 1H, J=1.2 Hz, =CH), 5.99 (d, 1H, J=0.9 Hz, =CH), 6.49 (s, 1H, =CH), 6.51 (s, 1H, =CH), 7.29-7.37 (m, 10H, 2×5ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =14.2, 14.3, 42.3, 42.6, 47.0, 47.6, 52.6, 52.7, 61.8, 63.3, 113.5, 116.0, 127.7, 128.5, 128.6, 128.9, 129.2, 136.9, 137.4, 139.0, 139.3, 163.4, 165.2, 166.4, 166.5; mass (FAB+) m/z 288 (M⁺+1); Anal. Calcd for C₁₆H₁₇NO₄: C, 66.89; H, 5.96; N, 4.88. Found: C, 67.08; H, 5.76; N, 4.98.

4.2.2. 3-(2-Chloro-phenyl)-2-cyano-4-methylene-pentanedioic acid-1-ethyl ester-5-methyl ester (4b). Yield 84% (2.3 g from 2.3 g) as a colorless oil; ν_{max} (neat) 1746 (CO₂Me and CO₂Et), 2253 (CN) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =1.13 (t, 3H, *J*=7.1 Hz, *CH*₃CH₂), 1.24 (t, 3H, *J*=7.1 Hz, *CH*₃CH₂), 3.69 (s, 3H, CO₂CH₃), 3.80 (s, 3H, CO₂CH₃), 4.13–4.25 (m, 4H, 2×*CH*₂CH₃), 4.33 (d, 1H, *J*=7.2 Hz, CHAr), 4.61 (d, 1H, *J*=7.2 Hz, CHAr), 5.08– 5.02 (m, 2H, 2×CHCN), 5.61 (d, 1H, *J*=0.9 Hz, ==CH), 6.15 (s, 1H, ==CH), 5.50 (s, 1H, ==CH), 6.60 (s, 1H, ==CH), 7.28–7.63 (m, 8H, 2×4ArH); mass (ES+) *m*/*z* 322.0 (M⁺+1); Anal. Calcd for C₁₆H₁₆ClNO₄: C, 59.73; H, 5.01; N, 4.35. Found: C, 59.79; H, 4.87; N, 4.51.

4.2.3. 2-Cyano-3-(2-fluoro-phenyl)-4-methylene-pentanedioic acid-1-ethyl ester-5-methyl ester (4c). Yield 92% (1.1 g from 1.0 g) as a pale yellow oil; ν_{max} (neat) 1744 (CO₂Me and CO₂Et), 2259 (CN) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =1.13–1.33 (m, 6H, 2×CH₃CH₂), 3.71 (s, 3H, CO₂CH₃), 3.78 (s, 3H, CO₂CH₃), 4.13–4.23 (m, 5H, 2×CH₂CH₃ and CHAr), 4.39 (d, 1H, *J*=7.3 Hz, CHAr), 4.93 (d, 2H, *J*=7.1 Hz, 2×CHCN), 5.75 (s, 1H, =CH), 6.00 (s, 1H, =CH), 6.50 (s, 1H, =CH), 6.58 (s, 1H, =CH), 7.11–7.30 (m, 6H, 2×3ArH), 7.53–7.56 (m, 2H, 2×1ArH); mass (ES+) *m*/*z* 306.1 (M⁺+1); Anal. Calcd for C₁₆H₁₆FNO₄: C, 62.94; H, 5.28; N, 4.59. Found: C, 63.01; H, 5.22; N, 4.58.

4.2.4. 3-(**4**-Bromo-phenyl)-**2**-cyano-4-methylene-pentanedioic acid-1-ethyl ester-5-methyl ester (4d). Yield 82% (2.2 g from 2.3 g) as a colorless oil; ν_{max} (neat) 1748 (CO₂Me and CO₂Et), 2254 (CN) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =1.18 (t, 3H, *J*=7.1 Hz, CH₃CH₂), 1.32 (t, 3H, J=7.1 Hz, CH₃CH₂), 3.71 (s, 3H, CO₂CH₃), 3.75 (s, 3H, CO₂CH₃), 4.13–4.41 (m, 6H, $2 \times CH_2$ CH₃ and $2 \times CHAr$), 4.55–4.58 (m, 2H, $2 \times CHCN$), 5.75 (s, 1H, =CH), 6.00 (s, 1H, =CH), 6.50 (s, 1H, =CH), 6.52 (s, 1H, =CH), 7.15–7.27 (m, 4H, 2×2 ArH), 7.43–7.51 (m, 4H, 2×2 ArH); mass (FAB+) *m*/*z* 366 (M⁺+1); Anal. Calcd for C₁₆H₁₆BrNO₄: C, 52.48; H, 4.40; N, 3.82. Found: C, 52.63; H, 4.54; N, 3.77.

4.2.5. 3-(4-Chloro-phenyl)-2-cyano-4-methylene-pentanedioic acid-1-ethyl ester-5-methyl ester (4e). Yield 76% (1.8 g from 2.0 g) as a colorless oil; ν_{max} (neat) 1747 (CO₂Me and CO₂Et), 2255 (CN) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =1.18 (t, 3H, J=7.1 Hz, CH₃CH₂), 1.33 (t, 3H, J=7.1 Hz, CH₃CH₂), 3.71 (s, 3H, CO₂CH₃), 3.75 (s, 3H, CO₂CH₃), 4.09–4.30 (m, 5H, 2×CH₂CH₃ and CHAr), 4.39 (d, 1H, J=7.3 Hz, CHAr), 4.56–4.63 (m, 2H, 2×CHCN), 5.76 (d, 1H, J=1.1 Hz, =CH), 6.00 (s, 1H, =CH), 6.50 (s, 1H, =CH), 6.52 (s, 1H, =CH), 7.25–7.32 (m, 8H, 2×4ArH); mass (ES+) *m/z* 344 (M⁺+Na); Anal. Calcd for C₁₆H₁₆ClNO₄: C, 59.73; H, 5.01; N, 4.35. Found: C, 59.89; H, 4.78; N, 4.31.

4.2.6. 2-Cyano-3-(4-fluoro-phenyl)-4-methylene-pentanedioic acid-1-ethyl ester-5-methyl ester (4f). Yield 72% (1.3 g from 1.5 g) as a colorless oil; v_{max} (neat) 1746 (CO₂Me and CO₂Et), 2254 (CN) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =1.16 (t, 3H, J=7.1 Hz, CH₃CH₂), 1.33 (t, 3H, J=7.1 Hz, CH₃CH₂), 3.71 (s, 3H, CO₂CH₃), 3.76 (s, 3H, CO₂CH₃), 4.12–4.22 (m, 4H, 2×CH₂CH₃), 4.28 (d, 1H, J=7.1 Hz, CHAr), 4.39 (d, 1H, J=7.3 Hz, CHAr), 4.58-4.60 (m, 2H, 2×CHCN), 5.75 (s, 1H, =CH), 6.00 (s, 1H, =CH), 6.49 (s, 1H, =CH), 6.51 (s, 1H, =CH), 7.00-7.09 (m, 4H, 2×2ArH), 7.23–7.39 (m, 4H, 2×2ArH); ¹³C NMR $(CDCl_3, 50 \text{ MHz}) \delta = 14.2, 14.3, 42.3, 42.6, 46.3, 47.1, 52.7,$ 52.8, 63.4, 116.0, 116.4, 127.7, 128.7, 130.2, 130.4, 130.6, 130.8, 132.6, 133.1, 138.8, 139.2, 165.0, 165.3, 166.3, 166.4; mass (ES+) m/z 306.2 (M⁺+1); Anal. Calcd for C₁₆H₁₆FNO₄: C, 62.94; H, 5.28; N, 4.59. Found: C, 63.11; H, 5.41; N, 4.67.

4.2.7. 2-Cyano-4-methylene-3-*p*-tolyl-pentanedioic acid-**1-ethyl ester-5-methyl ester (4g).** Yield 78% (1.89 g from 2.0 g) as a colorless oil; ν_{max} (neat) 1742 (CO₂Me and CO₂Et), 2256 (CN) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =1.16 (t, 3H, *J*=7.2 Hz, *CH*₃CH₂), 1.29 (t, 3H, *J*=7.2 Hz, *CH*₃CH₂), 2.31 (s, 3H, ArCH₃), 2.32 (s, 3H, ArCH₃), 3.71 (s, 3H, CO₂CH₃), 3.75 (s, 3H, CO₂CH₃), 4.11–4.22 (m, 5H, 2×*CH*₂CH₃ and CHAr), 4.40 (d, 1H, *J*=7.6 Hz, CHAr), 4.57–4.61 (m, 2H, 2×CHCN), 5.75 (d, 1H, *J*=1.0 Hz, =CH), 5.97 (s, 1H, =CH), 6.47 (s, 1H, =CH), 6.49 (s, 1H, =CH), 7.09–7.23 (m, 8H, 2×4ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =14.2, 14.5, 21.5, 42.4, 42.7, 46.7, 47.4, 52.6, 52.7, 61.9, 63.3, 116.1, 127.4, 128.3, 128.8, 129.9, 133.7, 134.3, 138.3, 139.1, 139.4, 165.3, 166.5, 166.6; mass (ES+) *m*/*z* 302.1 (M⁺+1); Anal. Calcd for C₁₇H₁₉NO₄: C, 67.76; H, 6.36; N, 4.65. Found: C, 67.49; H, 6.51; N, 4.67.

4.3. General procedure for the synthesis of compounds 5a–g, as exemplified for compound 5a

A mixture of compound **4a** (1.2 g, 4.18 mmol) and Raney-Ni (0.3 g, wet) in MeOH (20 mL) was hydrogenated at 40 psi in

the hydrogenation assembly (Parr) at room temperature. After completion, the catalyst was filtered over a bed of Celite and the filtrate was evaporated in vacuo to yield the crude product. Purification via silica gel column chromatography (hexane:EtOAc, 40:60, v/v) gave 0.65 g (60%) of product **5a** as a white solid.

4.3.1. 5-Methyl-6-oxo-4-phenyl-piperidine-3-carboxylic acid ethyl ester (5a). Mp 102–104 °C; ν_{max} (KBr) 1661 (CONH), 1730 (CO₂Et), 3402 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =0.86 (t, 3H, J=7.1 Hz, CH₃CH₂), 1.00–1.18 (m, 9H, CH₃CH₂ and 2×CH₃CH), 2.43–2.61 (m, 1H, CHCH₃), 2.71–2.78 (m, 1H, CHCH₃), 2.81–3.00 (m, 1H, CHCH₂), 3.07–3.22 (m, 1H, CHCH₂), 3.31–3.64 (m, 6H, 2×CH₂NH and 2×CHAr), 3.83 (q, 2H, J=7.1 Hz, CH₂CH₃), 4.08 (q, 2H, J=7.1 Hz, CH₂CH₃), 6.27 (s, 2H, 2×NH), 7.17–7.36 (m, 10H, 2×5ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =13.6, 14.0, 14.4, 15.2, 38.9, 42.2, 42.5, 43.9, 44.0, 45.7, 47.4, 50.2, 61.0, 61.5, 127.04, 127.7, 128.3, 128.6, 129.0, 140.4, 140.9, 172.2, 172.7, 175.2, 176.4; mass (ES+) *m*/*z* 262.1 (M⁺+1); Anal. Calcd for C₁₅H₁₉NO₃: C, 68.94; H, 7.33; N, 5.36. Found: C, 69.08; H, 7.55; N, 5.57.

4.3.2. 4-(2-Chloro-phenyl)-5-methyl-6-oxo-piperidine-3carboxylic acid ethyl ester (5b). Yield 67% (0.98 g from 1.6 g) as a pale yellow oil; v_{max} (neat) 1667 (CONH), 1731 (CO_2Et) , 3222 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) $\delta = 0.93$ (t, 3H, J = 7.1 Hz, CH_3CH_2), 1.01–1.18 (m, 9H, CH_3CH_2 and $2 \times CH_3CH$), 2.80–2.98 (m, 1H, CHCH₃), 3.17-3.20 (m, 2H, CHCH₃ and CHCH₂), 3.59-3.60 (m, 1H, CHCH₂), 3.63–3.66 (m, 4H, 2×CH₂NH), 3.87–3.92 (m, 2H, $2 \times CHAr$), 4.09–4.24 (m, 4H, $2 \times CH_2CH_3$), 6.19 (s, 2H, 2×NH), 7.14–7.21 (m, 6H, 2×3ArH), 7.32–7.42 (m, 2H, 2×1ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =13.7, 14.0, 14.3, 14.6, 37.2, 37.6, 41.5, 41.8, 42.6, 42.8, 43.4, 43.5, 61.2, 61.6, 127.1, 127.7, 128.6, 129.0, 130.2, 130.4, 134.7, 135.0, 137.4, 138.8, 171.7, 172.3, 174.9, 176.0; mass (FAB+) m/z 296 (M⁺+1); Anal. Calcd for C₁₅H₁₈ClNO₃: C, 60.91; H, 6.13; N, 4.74. Found: 60.82; H, 5.94; N, 4.89.

4.3.3. 4-(2-Fluoro-phenyl)-5-methyl-6-oxo-piperidine-3carboxylic acid ethyl ester (5c). Yield 57% (0.47 g from 0.9 g) as a colorless oil; ν_{max} (neat) 1652 (CONH), 1723 (CO₂Et), 3436 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =0.90 (t, 3H, *J*=7.1 Hz, *CH*₃CH₂), 1.02–1.29 (m, 9H, *CH*₃CH₂ and 2×*CH*₃CH), 2.62–2.93 (m, 2H, 2×*CHC*H₃), 3.24–3.26 (m, 2H, 2×*CHC*H₂), 3.58–3.61 (m, 4H, 2×*CH*₂NH), 3.87–3.91 (m, 2H, 2×*CHAr*), 4.06–4.13 (m, 4H, 2×*CH*₂CH₃), 6.37 (s, 2H, 2×NH), 7.00–7.22 (m, 8H, 2×4ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =13.6, 14.0, 14.3, 15.2, 37.9, 38.5, 41.2, 42.8, 43.1, 43.9, 44.7, 45.8, 61.3, 61.5, 115.9, 116.4, 124.5, 124.7, 127.2, 127.5, 127.8, 128.0, 128.9, 129.2, 129.5, 130.1, 172.0, 172.4, 174.8, 176.0; mass (FAB+) *m*/*z* 280 (M⁺+1); Anal. Calcd for C₁₅H₁₈FNO₃: C, 64.50; H, 6.50; N, 5.01. Found: C, 64.55; H, 6.41; N, 5.09.

4.3.4. 4-(4-Bromo-phenyl)-5-methyl-6-oxo-piperidine-3carboxylic acid ethyl ester (5d). Yield 58% (0.7 g from 1.3 g) as a white solid, mp 126–128 °C; ν_{max} (KBr) 1667 (CONH), 1724 (CO₂Et), 3313 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =0.93 (t, 3H, J=7.1 Hz, CH₃CH₂), 1.00 (d, 3H, J=7.3 Hz, CH₃CH), 1.08 (d, 3H, J=6.9 Hz, CH₃CH), 1.17 (t, 3H, J=7.1 Hz, CH₃CH₂), 2.38–2.60 (m, 1H, CHCH₃), 2.68–2.82 (m, 1H, CHCH₃), 2.75–3.10 (m, 2H, 2×CHCH₂), 3.45–3.68 (m, 6H, 2×CH₂NH and 2×CHAr), 3.85 (q, 2H, J=7.1 Hz, CH₂CH₃), 4.10 (q, 2H, J=7.2 Hz, CH₂CH₃), 6.21 (s, 2H, 2×NH), 7.07–7.10 (m, 4H, 2×2ArH), 7.44–7.54 (m, 4H, 2×2ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =13.6, 14.2, 14.4, 15.2, 38.8, 42.2, 42.4, 44.0, 44.1, 45.3, 47.2, 49.7, 61.2, 61.7, 121.4, 121.5, 130.0, 130.3, 132.2, 139.6, 140.0, 171.9, 172.4, 174.7, 176.0; mass (FAB+) *m*/*z* 340 (M⁺+1); Anal. Calcd for C₁₅H₁₈BrNO₃: C, 52.96; H, 5.33; N, 4.12. Found: C, 52.58; H, 5.49; N, 3.97.

4.3.5. 4-(4-Chloro-phenyl)-5-methyl-6-oxo-piperidine-3carboxylic acid ethyl ester (5e). Yield 62% (0.45 g from 0.79 g) as a white solid, mp 118–120 °C; ν_{max} (KBr) 1665 (CONH), 1733 (CO₂Et), 3303 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =0.92 (t, 3H, J=7.1 Hz, CH₃CH₂), 1.00 (d, 3H, J=7.2 Hz, CH₃CH), 1.08 (d, 3H, J=7.0 Hz, CH₃CH), 1.17 (t, 3H, J=7.1 Hz, CH₃CH₂), 2.40–2.58 (m, 1H, CHCH₃), 2.66–2.78 (m, 1H, CHCH₃), 2.93–3.10 (m, 2H, $2 \times CHCH_2$), 3.57–3.69 (m, 6H, $2 \times CH_2NH$ and $2 \times CHAr$), 3.85 (q, 2H, J=7.1 Hz, CH₂CH₃), 4.09 (q, 2H, J=7.1 Hz, CH_2CH_3), 6.23 (s, 2H, 2×NH), 7.09–7.15 (m, 4H, 2×2 ArH), 7.29–7.33 (m, 4H, 2×2 ArH); ¹³C NMR $(CDCl_3, 50 \text{ MHz}) \delta = 13.5, 14.1, 14.4, 15.2, 38.9, 42.3,$ 42.5, 44.0, 44.2, 45.2, 47.3, 49.7, 61.2, 61.6, 129.1, 129.2, 129.6, 130.0, 133.3, 133.5, 139.1, 139.5, 171.9, 172.5, 174.7, 176.2; mass (FAB+) m/z 296 (M⁺+1); Anal. Calcd for C₁₅H₁₈ClNO₃: C, 60.91; H, 6.13; N, 4.74. Found: C, 61.15; H, 5.85; N, 4.88.

4.3.6. 4-(4-Fluoro-phenyl)-5-methyl-6-oxo-piperidine-3carboxylic acid ethyl ester (5f). Yield 54% (0.26 g from 0.53 g) as a yellow solid, mp 125–127 °C; ν_{max} (KBr) 1663 (CONH), 1729 (CO₂Et), 3407 (NH) cm⁻¹; ¹H NMR $(CDCl_3, 200 \text{ MHz}) \delta = 0.90 \text{ (t, 3H, } J = 7.1 \text{ Hz}, CH_3CH_2),$ 1.02–1.29 (m, 9H, CH_3CH_2 and $2 \times CH_3CH$), 2.38–2.76 (m, 2H, 2×CHCH₃), 2.80–3.18 (m, 2H, CHCH₂), 3.53– 3.63 (m, 6H, 2×CH₂NH and 2×CHAr), 3.85 (q, 2H, J=7.0 Hz, CH₂CH₃), 4.11 (q, 2H, J=7.1 Hz, CH₂CH₃), 6.15 (s, 2H, 2×NH), 6.96-7.06 (m, 4H, 2×2ArH), 7.11-7.20 (m, 4H, 2×2 ArH); ¹³C NMR (CDCl₃, 50 MHz) $\delta = 13.6, 14.1, 14.4, 15.1, 39.0, 42.5, 44.0, 44.2, 45.0, 47.5,$ 49.6, 61.1, 61.6, 115.6, 116.0, 129.7, 129.8, 130.0, 130.2, 136.2, 136.7, 159.8, 164.6, 172.0, 172.6, 174.8, 176.2; mass (FAB+) m/z 280 (M⁺+1); Anal. Calcd for C₁₅H₁₈FNO₃: C, 64.50; H, 6.50; N, 5.01. Found: C, 64.80; H, 6.36; N, 4.88.

4.3.7. 5-Methyl-6-oxo-4*-p***-tolyl-piperidine-3-carboxylic** acid ethyl ester (5g). Yield 56% (0.56 g from 1.1 g) as a white solid, mp 105–107 °C; ν_{max} (KBr) 1663 (CONH), 1726 (CO₂Et), 3233 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =0.89 (t, 3H, *J*=7.2 Hz, CH₃CH₂), 1.01 (d, 3H, *J*=7.2 Hz, CH₃CH), 1.08 (d, 3H, *J*=7.2 Hz, CH₃CH), 1.17 (t, 3H, *J*=7.1 Hz, CH₃CH₂), 2.32 (s, 6H, 2×ArCH₃), 2.46–2.58 (m, 1H, CHCH₃), 2.68–2.76 (m, 1H, CHCH₃), 2.89–3.11 (m, 2H, 2×CHCH₂), 3.55–3.63 (m, 6H, 2×CH₂NH and 2×CHAr), 3.84 (q, 2H, *J*=7.1 Hz, CH₂CH₃), 4.09 (q, 2H, *J*=7.1 Hz, CH₂CH₃), 6.10 (s, 2H, 2×NH), 7.04–7.14 (m, 8H, 2×4ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =13.6, 14.1, 14.4, 15.2, 21.4, 38.9, 41.6, 42.4, 44.0, 44.1, 45.4, 47.5, 49.9, 61.0, 61.5, 128.1, 128.5, 129.6, 137.0, 137.2, 137.3, 137.9, 171.8, 172.2, 172.8, 175.3, 176.5; mass (FAB+) *m*/*z* 276 (M⁺+1); Anal. Calcd for C₁₆H₂₁NO₃: C, 69.79; H, 7.69; N, 5.09. Found: C, 69.98; H, 7.54; N, 4.89.

4.4. General procedure for the synthesis of compounds 8a–g, as exemplified for compound 8a

A mixture of compound **5a** (0.51 g, 1.95 mmol) and PCl₅ (1.62 g, 7.8 mmol) in POCl₃ (4 mL) was refluxed for 3 h. Thereafter, the reaction mixture was poured into ice water, neutralized with NaHCO₃, and extracted with EtOAc (2×25 mL). The combined organic layer was washed with brine (50 mL), dried (Na₂SO₄), and the solvent was in vacuo. The crude product obtained was purified by silica gel column chromatography (hexane:EtOAc, 50:50, v/v) to give 0.39 g (68%) of chloro-derivative **8a** as a white solid.

4.4.1. 5-Chloro-5-methyl-6-oxo-4-phenyl-piperidine-3carboxylic acid ethyl ester (8a). Mp 142–144 °C; ν_{max} (KBr) 1678 (CONH), 1736 (CO₂Et), 3200 (NH) cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ =0.90 (t, 3H, *J*=7.1 Hz, *CH*₃CH₂), 1.03 (t, 3H, *J*=7.1 Hz, *CH*₃CH₂), 1.56 (s, 3H, CH₃), 1.61 (s, 3H, CH₃), 3.32–3.38 (m, 2H, 2×CHAr), 3.62–3.72 (m, 4H, 2×CH₂NH), 3.78–4.05 (m, 6H, 2×CHCH₂ and 2×CH₂CH₃), 6.36 (br s, 1H, NH), 6.74 (br s, 1H, NH), 7.24–7.40 (m, 10H, 2×5ArH);¹³C NMR (CDCl₃, 50 MHz) δ =14.1, 14.2, 26.0, 26.2, 42.4, 43.1, 43.9, 44.5, 52.9, 54.8, 61.4, 61.7, 67.8, 68.1, 128.2, 128.3, 128.4, 128.7, 130.0, 136.1, 138.0, 170.7, 171.3, 171.6, 172.0; mass (FAB+) *m*/*z* 296 (M⁺+1); Anal. Calcd for C₁₅H₁₈ClNO₃: C, 60.91; H, 6.13; N, 4.74. Found: C, 61.19; H, 5.87; N, 4.78.

4.4.2. 5-Chloro-4-(2-chloro-phenyl)-5-methyl-6-oxopiperidine-3-carboxylic acid ethyl ester (8b). Yield 65% (0.59 g from 0.81 g) as a white crystalline solid, mp 142-144 °C; ν_{max} (KBr) 1685 (CONH), 1735 (CO₂Et), 3250 (NH) cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ =0.93 (t, 3H, J=6.0 Hz, CH_3CH_2), 1.17 (t, 3H, J=6.0 Hz, CH_3CH_2), 1.58 (s, 3H, CH₃), 1.60 (s, 3H, CH₃), 3.26-3.40 (m, 2H, 2×CHAr), 3.50-3.61 (m, 4H, 2×CH₂N), 3.86-4.03 (m, 6H, $2 \times CH_2CH_3$ and $2 \times CHCO_2Et$), 6.18 (s, 1H, NH), 6.47 (s, 1H, NH), 7.20–7.30 (m, 6H, 2×3ArH), 7.42–7.46 (m, 2H, 2×1ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =14.2, 14.3, 25.9, 26.0, 41.5, 42.0, 45.6, 45.9, 49.2, 50.1, 61.5, 61.9, 67.0, 68.0, 126.8, 127.4, 127.6, 129.3, 129.4, 129.9, 130.6, 131.5, 135.7, 137.8, 171.2, 171.4, 171.7, 171.9; mass (ES+) m/z 330.0 (M⁺+1), 332.1 (M⁺+3); Anal. Calcd for C₁₅H₁₇Cl₂NO₃: C, 54.56; H, 5.19; N, 4.24. Found: C, 54.33; H, 5.11; N, 4.52.

4.4.3. 5-Chloro-4-(2-fluoro-phenyl)-5-methyl-6-oxopiperidine-3-carboxylic acid ethyl ester (8c). Yield 63% (1.39 g from 2.0 g) as a white solid, mp 144–146 °C; ν_{max} (KBr) 1679 (CONH), 1735 (CO₂Et), 3401 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =0.93 (t, 3H, *J*=7.1 Hz, *CH*₃CH₂), 1.06 (t, 3H, *J*=7.1 Hz, *CH*₃CH₂), 1.62 (s, 3H, CH₃), 1.65 (s, 3H, CH₃), 3.28–3.41 (m, 2H, 2×CHAr), 3.55–3.63 (m, 4H, 2×CH₂N), 3.88–4.07 (m, 6H, $2 \times CH_2CH_3$ and $2 \times CHCO_2Et$), 6.52 (s, 1H, NH), 6.83 (s, 1H, NH), 7.03–7.18 (m, 6H, $2 \times 3ArH$), 7.27–7.31 (m, 2H, $2 \times 1ArH$); mass (ES+) *m*/*z* 314.0 (M⁺+1); Anal. Calcd for C₁₅H₁₇ClFNO₃: C, 57.42; H, 5.46; N, 4.46. Found: C, 57.60; H, 5.64; N, 4.31.

4.4.4. 4-(4-Bromo-phenyl)-5-chloro-5-methyl-6-oxopiperidine-3-carboxylic acid ethyl ester (8d). Yield 60% (0.13 g from 0.2 g) as a white solid, mp 148–150 °C; ν_{max} (KBr) 1697 (CONH), 1726 (CO₂Et), 3439 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =0.93–1.09 (m, 6H, 2×CH₃CH₂), 1.55 (s, 3H, CH₃), 1.59 (s, 3H, CH₃), 3.28– 3.47 (m, 2H, 2×CHAr), 3.58–3.83 (m, 4H, 2×CHCO₂Et and 2×CHHN), 3.89–4.06 (m, 6H, 2×CH₂CH₃ and 2×CHHN), 6.32 (s, 1H, NH), 6.67 (s, 1H, NH), 7.13–7.23 (m, 4H, 2×2ArH), 7.45–7.49 (m, 4H, 2×2ArH); mass (ES+) *m*/*z* 374 (M⁺+1); Anal. Calcd for C₁₅H₁₇BrClNO₃: C, 48.09; H, 4.57; N, 3.74. Found: C, 47.85; H, 4.83; N, 3.78.

4.4.5. 5-Chloro-4-(4-chloro-phenyl)-5-methyl-6-oxopiperidine-3-carboxylic acid ethyl ester (8e). Yield 54% (1.2 g from 2.0 g) as a white solid, mp 158–160 °C; ν_{max} (KBr) 1690 (CONH), 1728 (CO₂Et), 3312 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =0.96 (t, 3H, J=7.2 Hz, CH_3CH_2), 1.05 (t, 3H, J=7.2 Hz, CH_3CH_2), 1.55 (s, 3H, CH₃), 1.59 (s, 3H, CH₃), 3.28–3.37 (m, 2H, 2×CHAr), 3.58-3.88 (m, 4H, 2×CHCO₂Et and 2×CHHN), 3.90-4.06 (m, 6H, $2 \times CH_2CH_3$ and $2 \times CHHN$), 6.45 (s, 1H, NH), 6.81 (s, 1H, NH), 7.19–7.34 (m, 8H, 2×4ArH); ¹³C NMR $(CDCl_3, 50 \text{ MHz}) \delta = 14.2, 14.3, 25.9, 26.1, 42.4, 43.2,$ 43.8, 44.3, 52.3, 54.2, 61.6, 61.8, 67.5, 67.8, 128.6, 128.9, 131.3, 131.8, 134.3, 134.5, 134.6, 136.2, 170.3, 170.9, 171.5, 171.8; mass (FAB+) m/z 330 (M⁺+1); Anal. Calcd for C₁₅H₁₇Cl₂NO₃: C, 54.56; H, 5.19; N, 4.24. Found: C, 54.77; H, 5.02; N, 4.31.

4.4.6. 5-Chloro-4-(4-fluoro-phenyl)-5-methyl-6-oxopiperidine-3-carboxylic acid ethyl ester (8f). Yield 72% (0.97 g from 1.2 g) as a white solid, mp 110–112 °C; ν_{max} (KBr) 1696 (CONH), 1728 (CO₂Et), $3\overline{4}12$ (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =0.94 (t, 3H, J=7.2 Hz, CH₃CH₂), 1.04 (t, 3H, J=7.2 Hz, CH₃CH₂), 1.56 (s, 3H, CH₃), 1.59 (s, 3H, CH₃), 3.29–3.44 (m, 2H, 2×CHAr), 3.57-3.67 (m, 4H, 2×CH₂N), 3.83-4.01 (m, 6H, 2×CH₂CH₃ and 2×CHCO₂Et), 6.36 (s, 1H, NH), 6.72 (s, 1H, NH), 6.99–7.07 (m, 4H, 2×2ArH), 7.31–7.34 (m, 4H, 2×2 ArH), ¹³C NMR (CDCl₃, 50 MHz) $\delta = 14.1$, 14.2, 25.9, 26.0, 42.5, 43.2, 43.8, 44.4, 52.2, 54.1, 61.5, 61.8, 67.7, 68.1, 115.1, 115.4, 115.5, 115.8, 131.6, 132.0, 133.4, 170.5, 171.0, 171.7, 171.9; mass (ES+) m/z 314.0 (M⁺+1); Anal. Calcd for C₁₅H₁₇ClFNO₃: C, 57.42; H, 5.46; N, 4.46. Found: C, 57.71; H, 5.40; N, 4.44.

4.4.7. 5-Chloro-5-methyl-6-oxo-4-*p*-tolyl-piperidine-3carboxylic acid ethyl ester (8g). Yield 67% (0.45 g from 0.6 g) as a white solid, mp 134–136 °C; ν_{max} (KBr) 1679 (CONH), 1724 (CO₂Et), 3365 (NH) cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ =0.94 (t, 3H, *J*=7.5 Hz, CH₃CH₂), 1.07 (t, 3H, *J*=7.5 Hz, CH₃CH₂), 1.56 (s, 3H, CH₃), 1.61 (s, 3H, CH₃), 2.34 (s, 6H, 2×ArCH₃), 3.23–3.40 (m, 2H, 2×CHAr), 3.78–3.80 (m, 3H, 2×CHCO₂Et and CHHN), 3.84–4.06 (m, 7H, 2×CH₂CH₃, CHHN and 2×CHHN), 6.18 (s, 1H, NH), 6.56 (s, 1H, NH), 7.12–7.14 (m, 4H, 2×2ArH), 7.24–7.27 (m, 4H, 2×2ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =14.1, 14.2, 21.6, 26.0, 26.2, 42.4, 43.0, 43.9, 44.5, 52.5, 54.4, 61.4, 61.6, 68.0, 68.2, 129.0, 129.4, 129.8, 130.4, 133.0, 135.0, 137.9, 138.1, 170.8, 171.3, 171.9, 172.0; mass (ES+) *m*/*z* 310 (M⁺+1), 332.0 (M⁺+Na); Anal. Calcd for C₁₆H₂₀ClNO₃: C, 62.03; H, 6.51; N, 4.52. Found: C, 61.86; H, 6.42; N, 4.60.

4.5. General procedure for the synthesis of compounds 9a–g, as exemplified for compound 9a

To the solution of compound 8a (0.5 g, 1.69 mmol) in anhydrous acetonitrile was added DBU (0.52 mL, 3.39 mmol) and the reaction mixture was heated at reflux for 14 h. Thereafter, the solvent was evaporated in vacuo to yield a residue, which via silica gel column chromatography (hexane: EtOAc, 40:60 v/v) afforded 0.28 g (64%) of compound **9a** as a white solid.

4.5.1. 5-Methyl-4-oxo-6-phenyl-3-aza-bicyclo[3.1.0]hexane-1-carboxylic acid ethyl ester (9a). Mp 115–117 °C; ν_{max} (KBr) 1678 (CONH), 1718 (CO₂Et), 3413 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =1.02 (t, 3H, *J*=7.2 Hz, CH₃CH₂), 1.52 (s, 3H, CH₃), 2.66 (s, 1H, CHAr), 3.66 (d, 1H, *J*=10.4 Hz, CHHN), 3.90 (d, 1H, *J*=10.6 Hz, CHHNH), 4.05 (q, 2H, *J*=7.2 Hz, CH₂CH₃), 5.98 (s, 1H, NH), 7.15–7.28 (m, 3H, ArH), 7.32–7.40 (m, 2H, ArH); ¹³C NMR (CDCl₃, 75 MHz) δ =9.1, 14.1, 35.0, 35.4, 37.9, 45.1, 61.3, 128.2, 128.4, 129.6, 136.0, 170.1, 178; mass (ES+) *m*/*z* 260.2 (M⁺+1); Anal. Calcd for C₁₅H₁₇NO₃: C, 69.48; H, 6.61; N, 5.40. Found: C, 69.42; H, 6.55; N, 5.31.

4.5.2. 6-(2-Chloro-phenyl)-5-methyl-4-oxo-3-aza-bicyclo[3.1.0]hexane-1-carboxylic acid ethyl ester (9b). Yield 61% (0.27 g from 0.5 g) as a yellow oil; ν_{max} (neat) 1679 (CONH), 1732 (CO₂Et), 3406 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =0.99 (t, 3H, *J*=7.2 Hz, *CH*₃CH₂), 1.60 (s, 3H, CH₃), 2.49 (s, 1H, CHAr), 3.75 (d, 1H, *J*=10.5 Hz, *CH*HNH), 3.93–4.07 (m, 3H, CH*H*NH and *CH*₂CH₃), 6.11 (s, 1H, NH), 7.21–7.32 (m, 3H, ArH), 7.34–7.40 (m, 1H, ArH); ¹³C NMR (CDCl₃, 75 MHz) δ =9.3, 14.2, 34.5, 35.6, 37.9, 45.2, 52.2, 61.3, 127.2, 128.6, 130.2, 134.6, 137.4, 171.2, 175.4; mass (ES+) *m/z* 294.1 (M⁺+1); Anal. Calcd for C₁₅H₁₆ClNO₃: C, 61.31; H, 5.49; N, 4.77. Found: C, 61.59; H, 5.70; N, 4.68.

4.5.3. 6-(2-Fluoro-phenyl)-5-methyl-4-oxo-3-aza-bicy-clo[3.1.0]hexane-1-carboxylic acid ethyl ester (9c). Yield 65% (0.16 g from 0.28 g) as a white solid, mp 142–144 °C; ν_{max} (KBr) 1657 (CONH), 1729 (CO₂Et), 3280 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =1.04 (t, 3H, *J*=7.2 Hz, *CH*₃CH₂), 1.52 (s, 3H, CH₃), 2.48 (s, 1H, CHAr), 3.65 (d, 1H, *J*=10.8 Hz, *CH*HN), 3.97–4.09 (m, 3H, *CH*₂CH₃ and CH*H*N), 5.95 (s, 1H, NH), 6.98–7.15 (m, 2H, ArH), 7.23–7.27 (m, 2H, ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =9.2, 14.2, 34.4, 35.4, 38.0, 45.3, 61.3, 115.4, 115.8, 121.0, 121.3, 124.2, 129.4, 129.6, 131.8, 169.1, 178.5; mass (FAB+) *m*/*z* 278 (M⁺+1); Anal. Calcd for C₁₅H₁₆FNO₃: C, 64.57; H, 5.82; N, 5.05. Found: C, 64.77; H, 5.70; N, 4.86.

4.5.4. 6-(4-Bromo-phenyl)-5-methyl-4-oxo-3-aza-bicyclo[3.1.0]hexane-1-carboxylic acid ethyl ester (9d). Yield

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58% (0.05 g from 0.1 g) as a white solid, mp 136–138 °C; ν_{max} (KBr) 1704 (CONH), 1728 (CO₂Et), 3268 (NH) cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ =1.08 (t, 3H, J=7.5 Hz, CH₃CH₂), 1.51 (s, 3H, CH₃), 2.59 (s, 1H, CHAr), 3.66 (d, 1H, J=12.0 Hz, CHHN), 3.93 (d, 1H, J=12.0 Hz, CHHN), 4.08 (q, 2H, J=7.5 Hz, CH₂CH₃), 5.72 (s, 1H, NH), 7.07 (d, 2H, J=8.5 Hz, ArH), 7.45 (d, 2H, J=8.5 Hz, ArH); ¹³C NMR (CDCl₃, 75 MHz) δ =8.8, 13.9, 32.2, 37.4, 38.5, 44.8, 61.1, 121.4, 131.4, 131.5, 132.0, 168.8, 177.7; mass (ES+) *m*/*z* 338.1 (M⁺+1), 340.1 (M⁺+3); Anal. Calcd for C₁₅H₁₆BrNO₃: C, 53.27; H, 4.77; N, 4.14. Found: C, 53.55; H, 4.91; N, 3.86.

4.5.5. 6-(4-Chloro-phenyl)-5-methyl-4-oxo-3-aza-bicyclo[3.1.0]hexane-1-carboxylic acid ethyl ester (9e). Yield 60% (0.16 g from 0.3 g) as a white crystalline solid, mp 125–127 °C; ν_{max} (KBr) 1705 (CONH), 1732 (CO₂Et), 3312 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =1.07 (t, 3H, *J*=7.1 Hz, *CH*₃CH₂), 1.50 (s, 3H, CH₃), 2.60 (s, 1H, CHAr), 3.65 (d, 1H, *J*=10.0 Hz, *CH*HN), 3.98 (d, 1H, *J*=10.5 Hz, CH*H*N), 4.00 (q, 2H, *J*=7.1 Hz, *CH*₂CH₃), 5.69 (s, 1H, NH), 7.10 (d, 2H, *J*=8.3 Hz, ArH), 7.29 (d, 2H, *J*=8.2 Hz, ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =9.2, 14.3, 35.7, 37.9, 38.8, 45.4, 61.4, 128.9, 131.6, 131.8, 133.6, 169.3, 178.3; mass (FAB+) *m*/*z* 294 (M⁺+1); Anal. Calcd for C₁₅H₁₆ClNO₃: C, 61.33; H, 5.49; N, 4.77. Found: C, 61.68; H, 5.41; N, 4.49.

4.5.6. 6-(**4**-Fluoro-phenyl)-5-methyl-4-oxo-3-aza-bicyclo[3.1.0]hexane-1-carboxylic acid ethyl ester (9f). Yield 63% (0.14 g from 0.25 g) as a yellow oil; ν_{max} (neat) 1701 (CONH), 1738 (CO₂Et), 3400 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =1.06 (t, 3H, *J*=7.2 Hz, C*H*₃CH₂), 1.50 (s, 3H, CH₃), 2.61 (s, 1H, CHAr), 3.65 (d, 1H, *J*=11.2 Hz, C*H*HNH), 3.92 (d, 1H, *J*=11.0 Hz, CH*H*NH), 3.98–4.13 (m, 2H, C*H*₂CH₃), 5.89 (s, 1H, NH), 6.98–7.22 (m, 4H, ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =9.2, 14.3, 35.7, 38.2, 38.8, 45.6, 61.4, 115.4, 115.8, 128.2, 129.2, 131.7, 131.9, 169.4, 178.9; mass (ES+) *m*/*z* 278.1 (M⁺+1); Anal. Calcd for C₁₅H₁₆FNO₃: C, 64.97; H, 5.82; N, 5.05. Found: C, 64.90; H, 5.69; N, 4.98.

4.5.7. 5-Methyl-4-oxo-6-*p*-tolyl-3-aza-bicyclo[3.1.0] hexane-1-carboxylic acid ethyl ester (9g). Yield 65% (0.23 g from 0.4 g) as a white solid, mp 156–158 °C; ν_{max} (KBr) 1696 (CONH), 1735 (CO₂Et), 3426 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =1.05 (t, 3H, *J*=7.1 Hz, *CH*₃CH₂), 1.50 (s, 3H, CH₃), 2.33 (s, 3H, ArCH₃), 2.62 (s, 1H, CHAr), 3.65 (d, 1H, *J*=10.5 Hz, *CH*HN), 3.89 (d, 1H, *J*=10.4 Hz, CH*H*N), 4.08 (q, 2H, *J*=7.1 Hz, *CH*₂CH₃), 5.80 (s, 1H, NH), 7.04 (d, 2H, *J*=8.1 Hz, ArH), 7.12 (d, 2H, *J*=8.2 Hz, ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =9.3, 14.3, 21.6, 35.6, 38.0, 39.5, 42.6, 61.8, 128.5, 129.4, 136.7, 139.6, 168.0, 171.2; mass (ES+) *m*/*z* 274.1 (M⁺+1); Anal. Calcd for C₁₆H₁₉NO₃: C, 70.31; H, 7.01; N, 5.12. Found: C, 70.10; H, 6.75; N, 4.88.

4.6. General procedure for the synthesis of compounds **10b,e,g, as exemplified for compound 10g**

Compound **4g** (0.3 g, 1.0 mmol) and FeCl₃· $6H_2O$ (0.80 g, 3.0 mmol) were dissolved in propionic acid (6 mL) and the mixture was heated at reflux for 2 h. After cooling the

mixture to room temperature, it was poured into 1 N HCl (20 mL) and extracted with EtOAc (3×30 mL). The organic layers were pooled and washed with NaHCO₃ (50 mL), dried over Na₂SO₄, and evaporated to yield a residue, which was purified via silica gel column chromatography. Elution with hexane:EtOAc (60:40, v/v) yielded 0.13 g (60%) of product **10g** as a brown solid.

4.6.1. 4-(2-Chloro-phenyl)-3-methylene-piperidine-2,6dione (10b). Yield 58% (0.26 g from 0.62 g) as a white solid, 108–110 °C; ν_{max} (KBr) 1701 (CONH), 3408 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =2.94–3.05 (m, 2H, CH₂CH), 4.54–4.59 (m, 1H, CHAr), 5.33 (s, 1H, =CH), 6.52 (d, 1H, *J*=0.9 Hz, =CH), 7.15–7.32 (m, 3H, ArH), 7.43–7.48 (m, 1H, ArH), 8.24 (s, 1H, NH);¹³C NMR (CDCl₃, 50 MHz) δ =37.7, 39.1, 127.2, 127.9, 128.5, 129.4, 130.6, 137.1, 137.7, 166.2, 171.8; mass (FAB+) *m/z* 236 (M⁺+1); Anal. Calcd for C₁₂H₁₀ClNO₂: C, 61.16; H, 4.28; N, 5.94. Found: C, 60.80; H, 4.49; N, 6.01.

4.6.2. 4-(4-Chloro-phenyl)-3-methylene-piperidine-2,6dione (10e). Yield 65% (0.23 g from 0.49 g) as a white solid, 182–184 °C; ν_{max} (KBr) 1699 (CONH), 3404 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =2.96–2.99 (m, 2H, CH₂CH), 4.02–4.12 (m, 1H, CHAr), 5.39 (s, 1H, ==CH), 6.50 (s, 1H, ==CH), 7.15 (d, 2H, *J*=8.5 Hz, ArH), 7.36 (d, 2H, *J*=8.5 Hz, ArH), 8.19 (br s, 1H, NH);¹³C NMR (CDCl₃, 50 MHz) δ =43.3, 46.5, 131.2, 134.2, 134.4, 138.1, 143.3, 144.5, 171.1, 177.0; mass (FAB+) *m/z* 236 (M⁺+1); Anal. Calcd for C₁₂H₁₀CINO₂: C, 61.16; H, 4.28; N, 5.94. Found: C, 61.22; H, 4.03; N, 5.78.

4.6.3. 3-Methylene-4*-p***-tolyl-piperidine-2,6-dione (10g).** Mp 152–154 °C; ν_{max} (KBr) 1697 (CONH), 3427 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =2.35 (s, 3H, ArCH₃), 2.89–3.02 (m, 2H, *CH*₂CH), 3.99–4.15 (m, 1H, CHAr), 5.39 (s, 1H, =CH), 6.47 (s, 1H, =CH), 7.09 (d, 2H, *J*=8.0 Hz, ArH), 7.19 (d, 2H, *J*=8.0 Hz, ArH), 8.18 (s, 1H, NH); mass (FAB+) *m/z* 216 (M⁺+1); Anal. Calcd for C₁₃H₁₃NO₂: C, 72.54; H, 6.09; N, 6.51. Found: C, 72.67; H, 5.85; N, 6.82.

4.7. General procedure for the synthesis of compounds 11a,e,f, as exemplified for compound 11a

To a flask charged with compound **4a** (0.9 g, 3.14 mmol) was added 6 mL mixture of TFA and H_2SO_4 (1: 1) at room temperature and the reaction was continued for 4 h. Thereafter, the reaction mixture was poured into ice cold water (50 mL) and neutralized with NaHCO₃ and extracted with EtOAc (2×30 mL). Combined organic layer was dried over Na₂SO₄ and evaporated in vacuo to afford a residue, which was purified via column chromatography over silica gel using hexane:EtOAc (30:70, v/v) to furnish 0.76 g (80%) of amide **11a** as a white solid.

4.7.1. 2-Carbamoyl-4-methylene-3-phenyl-pentanedioic acid-1-ethyl ester-5-methyl ester (11a). Mp 120–122 °C; ν_{max} (KBr) 1666 (CONH₂), 1724 (CO₂Me and CO₂Et), 3396 (NH₂) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =1.01 (t, 3H, *J*=7.1 Hz, CH₃CH₂), 1.26 (t, 3H, *J*=7.1 Hz, CH₃CH₂), 3.69 (s, 6H, 2×CO₂CH₃), 3.97 (q, 2H, *J*=7.1 Hz, CH₂CH₃), 4.10–4.21 (m, 4H, CH₂CH₃ and 2×CHAr), 4.53 (d, 1H, *J*=12.4 Hz, CHCON), 4.62 (d, 1H, *J*=12.4 Hz, CHCON), 5.36 (s, 1H, 1H of NH₂), 5.53 (s, 1H, 1H of NH₂), 5.80 (s, 1H, =CH), 5.94 (s, 1H, =CH), 5.98 (s, 1H, 1H of NH₂), 6.28 (s, 1H, 1H of NH₂), 6.32 (s, 1H, =CH), 6.33 (s, 1H, =CH), 7.14–7.20 (m, 6H, 2×3ArH), 7.38–7.42 (m, 4H, 2×2ArH); ¹³C NMR (CDCl₃, 50 MHz) δ =19.0, 19.3, 50.7, 50.8, 57.1, 61.5, 61.7, 66.4, 66.5, 125.9, 129.6, 130.5, 135.6, 135.9, 136.3, 143.6, 144.0, 145.3, 146.5, 171.4, 173.5, 173.8; mass (FAB+) *m*/*z* 306 (M⁺+1); Anal. Calcd for C₁₆H₁₉NO₅: C, 62.94; H, 6.27; N, 4.59. Found: C, 63.05; H, 6.13; N, 4.83.

4.7.2. 2-Carbamovl-3-(4-chloro-phenvl)-4-methylenepentanedioic acid-1-ethyl ester-5-methyl ester (11e). Yield 89% (0.63 g from 0.67 g) as a white solid, mp 102-104 °C; ν_{max} (KBr) 1667 (CONH₂), 1725 (CO₂Me and CO₂Et), 3389 (NH₂) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) $\delta = 1.02$ (t, 3H, J = 7.2 Hz, CH_3CH_2), 1.25 (t, 3H, J=7.2 Hz, CH_3CH_2), 3.69 (s, 6H, $2 \times CO_2CH_3$), 3.96 (q, 2H, J=7.2 Hz, CH₂CH₃), 4.11-4.25 (m, 4H, CH₂CH₃ and 2×CHAr), 4.54 (d, 1H, J=12.4 Hz, CHCON), 4.63 (d, 1H, J=12.4 Hz, CHCON), 5.62 (s, 1H, 1H of NH₂), 5.73 (s, 1H, 1H of NH₂), 5.81 (s, 1H, =CH), 5.85 (s, 1H, 1H of NH₂), 5.94 (s, 1H, =CH), 5.98 (s, 1H, 1H of NH₂), 6.32 (s, 1H, =CH), 6.33 (s, 1H, =CH), 7.23–7.31 (m, 8H, 2×4 ArH); ¹³C NMR (CDCl₃, 50 MHz) $\delta = 19.0$, 19.2, 50.8, 51.1, 57.3, 61.7, 61.8, 66.3, 66.5, 129.5, 130.5, 133.4, 135.2, 135.4, 137.7, 142.9, 143.4, 145.4, 146.5, 171.4, 173.6, 173.8; mass (ES+) m/z 340.0 (M++1); 342.0 (M⁺+3); Anal. Calcd for C₁₆H₁₈ClNO₅: C, 56.56; H, 5.34; N, 4.12. Found: C, 56.79; H, 5.22; N, 3.95.

4.7.3. 2-Carbamoyl-3-(4-fluoro-phenyl)-4-methylenepentanedioic acid-1-ethyl ester-5-methyl ester (11f). Yield 78% (0.68 g from 0.83 g) as a white solid, mp 110-112 °C; ν_{max} (KBr) 1664 (CONH₂), 1726 (CO₂Me and CO_2Et), 3403 (NH₂) cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) $\delta = 1.02$ (t, 3H, J=7.5 Hz, CH₃CH₂), 1.27 (t, 3H, J=7.5 Hz, CH₃CH₂), 3.70 (s, 3H, CO₂CH₃), 3.71 (s, 3H, CO₂CH₃), 3.94–4.01 (m, 3H, CH₂CH₃ and CHAr), 4.12– 4.25 (m, 3H, CH₂CH₃ and CHAr), 4.57 (d, 1H, J=12.0 Hz, CHCON), 4.66 (d, 1H, J=12.0 Hz, CHCON), 5.28 (s, 1H, 1H of NH₂), 5.43 (s, 1H, 1H of NH₂), 5.81 (s, 1H, =CH), 5.95 (s, 2H, =CH and 1H of NH₂), 6.28 (s, 1H, 1H of NH₂), 6.34 (s, 2H, 2×=CH), 6.94-7.00 (m, 4H, 2×2ArH), 7.26–7.32 (m, 4H, 2×2ArH); ¹³C NMR $(CDCl_3, 50 \text{ MHz}) \delta = 14.1, 14.3, 46.1, 46.6, 52.3, 57.5,$ 61.7, 61.8, 115.1, 115.6, 124.5, 125.7, 130.3, 130.5, 130.6, 134.9, 135.2, 140.5, 141.7, 159.6, 164.4, 166.6, 166.7, 168.6, 169.2; mass (ES+) m/z 324.1 (M⁺+1); Anal. Calcd for C₁₆H₁₈FNO₅: C, 59.44; H, 5.61; N, 4.33. Found: C, 59.63; H, 5.40; N, 4.47.

4.8. General procedure for the synthesis of compounds 12a,e,f, as exemplified for compound 12a

To the stirred solution of compound **11a** (0.5 g, 1.64 mmol) in anhydrous toluene was added NaH (0.098 g in 60% oil, 2.46 mmol) at ambient temperature. After 30 min reaction mixture was quenched carefully with water and extracted with ethyl acetate (2×25 mL). The organic layer was dried (Na₂SO₄) and evaporated in vacuo to obtain a residue that was subjected to column chromatography using

hexane:EtOAc (40:60 v/v) over silica gel to yield 0.32 g (72%) of compound **12a** as a white solid.

4.8.1. 5-Methylene-2,6-dioxo-4-phenyl-piperidine-3-carboxylic acid ethyl ester (12a). Mp 114–116 °C; ν_{max} (KBr) 1705 (CONH), 1748 (CO₂Et), 3418 (NH) cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ =1.13 (t, 3H, J=7.5 Hz, CH₃CH₂), 3.40 (d, 1H, J=9.0 Hz, CHAr), 4.14 (q, 2H, J=7.5 Hz, CH₂CH), 4.43 (d, 1H, J=9.0 Hz, CHCO₂Et), 5.45 (s, 1H, =CH), 6.57 (s, 1H, =CH), 7.21–7.33 (m, 3H, ArH), 7.35–7.41 (m, 2H, ArH), 8.15 (s, 1H, NH); ¹³C NMR (CDCl₃, 50 MHz) δ =14.3, 46.2, 55.3, 62.5, 128.3, 128.6, 129.6, 137.2, 137.4, 165.1, 167.5, 168.3; mass (ES+) *m*/*z* 274.1 (M⁺+1); Anal. Calcd for C₁₅H₁₅NO₄: C, 65.52; H, 5.53; N, 5.13. Found: C, 65.59; H, 5.71; N, 4.93.

4.8.2. 4-(4-Chloro-phenyl)-5-methylene-2,6-dioxo-piperidine-3-carboxylic acid ethyl ester (12e). Yield 74% (0.15 g from 0.22 g) as a white solid, mp 108–110 °C; ν_{max} (KBr) 1695 (CONH), 1748 (CO₂Et), 3418 (NH) cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ =1.14 (t, 3H, *J*=7.1 Hz, CH₃CH₂), 3.40 (d, 1H, *J*=9.5 Hz, CHAr), 4.14 (q, 2H, *J*=7.1 Hz, CH₂CH), 4.38 (d, 1H, *J*=9.2 Hz, CHCO₂Et), 5.40 (s, 1H, =CH), 6.58 (s, 1H, =CH), 7.16 (d, 2H, *J*=7.8 Hz, ArH), 7.36 (d, 2H, *J*=8.0 Hz, ArH), 8.21 (s, 1H, NH); ¹³C NMR (CDCl₃, 50 MHz) δ =14.3, 45.5, 51.1, 62.6, 129.3, 130.5, 134.5, 135.7, 137.1, 164.8, 167.2, 167.9; mass (FAB+) *m*/*z* 308 (M⁺+1); Anal. Calcd for C₁₅H₁₄ClNO₄: C, 58.54; H, 4.59; N, 4.55. Found: C, 58.50; H, 4.69; N, 4.58.

4.8.3. 4-(**4**-Fluoro-phenyl)-5-methylene-2,6-dioxo-piperidine-3-carboxylic acid ethyl ester (12f). Yield 76% (0.25 g from 0.36 g) as a white solid, mp 135–137 °C; ν_{max} (KBr) 1703 (CONH), 1740 (CO₂Et), 3372 (NH) cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ =1.15 (t, 3H, *J*=7.5 Hz, CH₃CH₂), 3.95 (d, 1H, *J*=9.0 Hz, CHAr), 4.15 (q, 2H, *J*=7.5 Hz, CH₂CH), 4.40 (d, 1H, *J*=9.0 Hz, CHCO₂Et), 5.41 (s, 1H, =CH), 6.58 (s, 1H, =CH), 7.06–7.12 (m, 2H, ArH), 7.19–7.24 (m, 2H, ArH), 8.12 (s, 1H, NH); ¹³C NMR (CDCl₃, 50 MHz) δ =14.3, 45.4, 55.4, 62.6, 116.3, 116.8, 129.6, 130.0, 130.2, 132.9, 137.4, 165.0, 167.3, 168.1; mass (FAB+) *m*/*z* 292 (M⁺+1); Anal. Calcd for C₁₅H₁₄FNO₄: C, 61.58; H, 4.85; N, 4.81. Found: C, 61.55; H, 4.56; N, 4.77.

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References and notes

 (a) Basavaiah, D.; Jaganmohan Rao, A.; Satyanarayana, T. *Chem. Rev.* 2003, 811; (b) Basavaiah, D.; Reddy, R. J.; Rao, J. S. *Tetrahedron Lett.* 2006, 47, 73; (c) Coelho, F.; Veronese, D.; Pavam, C. H.; de Paula, V. I.; Buffon, R. *Tetrahedron* 2006, 62, 4563; (d) Chandrasekhar, S.; Basu, D.; Rambabu, Ch. *Tetrahedron Lett.* 2006, 47, 3059; (e) Shi, Y.-L.; Shi, M. *Synlett* 2005, 2623; (f) Lee, K. Y.; Gowrisankar, S.; Kim, J. N. Bull. Korean Chem. Soc. 2005, 26, 1481; (g) Shanmugam,
P.; Rajasingh, P. Tetrahedron Lett. 2005, 46, 3369; (h) Mix, S.;
Blechert, S. Org. Lett. 2005, 7, 2015; (i) Du, Y.; Feng, J.; Lu, X.
Org. Lett. 2005, 7, 1987; (j) Nair, Vijay; Abhilash, K. G.
Synthesis 2005, 12, 1967.

- (a) Singh, V.; Batra, S. Synthesis 2006, 63; (b) Singh, V.; Saxena, R.; Batra, S. J. Org. Chem. 2005, 70, 353; (c) Pathak, R.; Roy, A. K.; Batra, S. Synlett 2005, 848; (d) Pathak, R.; Roy, A. K.; Kanojiya, S.; Batra, S. Tetrahedron Lett. 2005, 46, 5289; (e) Batra, S.; Roy, A. K. Synthesis 2004, 2550; (f) Nag, S.; Pathak, R.; Kumar, M.; Shukla, P. K.; Batra, S. Bioorg. Med. Chem. Lett. 2006, 16, 3824.
- Weintraub, P. M.; Sabol, J. S.; Kane, J. M.; Borcherding, D. R. *Tetrahedron* 2003, 59, 2953.
- 4. (a) Kumar, S.; Raje, N.; Hideshima, T.; Ishitsuka, K.; Podar, K.; Le Gouille, S.; Chauhan, D.; Richardson, P.; Munshi, N. C.; Anderson, K. Br. J. Haematol. 2006, 132, 698; (b) Gaul, C.; Njardarson, J. T.; Danishefsky, S. J. J. Am. Chem. Soc. 2003, 125, 6042; (c) Gaul, C.; Njardarson, J. T.; Shan, D.; Dorn, D. C.; Wu, K.-D.; Tong, W. P.; Huang, X.-Y.; Moore, M. A. S.; Danishefsky, S. J. J. Am. Chem. Soc. 2004, 126, 11326; (d) Powell, R. G.; Smith, C. R., Jr.; Weisleder, D. J. Am. Chem. Soc. 1983, 105, 3739; (e) Suarez, A. I.; Blanco, Z.; Monache, F. D.; Compagnone, R. S.; Arvelo, F. Nat. Prod. Res. 2004, 18, 421; (f) Nakae, K.; Yoshimoto, Y.; Sawa, T.; Homma, Y.; Hamada, M.; Takeuchi, T.; Imoto, M. J. Antibiot. 2000, 53, 1130; (g) Urakawa, A.: Otani, T.: Yoshida, K.-I.; Nakayama, M.; Suzukake-Tsuchiya, K.; Hori, M. J. Antibiot. 1993, 46, 1827; (h) Ahmed, S. J. Mol. Struct. (Theochem) 1998, 422, 271.

- (a) Baylis, A. B.; Hillman, M. E. D. German Patent 2155113, 1972; *Chem. Abstr.* 1972, 77, 34174q; (b) Patra, A.; Batra, S.; Kundu, B.; Joshi, B. S.; Roy, R.; Bhaduri, A. P. *Synthesis* 2001, 276.
- Thomas, O. P.; Dumas, C.; Zaparucha, A.; Husson, H. P. *Eur. J. Org. Chem.* 2004, 1128.
- 7. Prugh, J. D.; Deana, A. A.; Wiggins, J. Synthesis 1989, 554.
- 8. Crystal data of compound 9e: C₁₅H₁₆NO₃Cl, M=293.75, orthorhombic, $P2_{1}2_{1}2_{1}$, a=6.323(1),b = 9.076(2),c=26.253(2) Å, V=1506.5(4) Å³, Z=4, $D_c=1.295$ g cm⁻³, μ (Mo K_{σ})=0.026 mm⁻¹, F(000)=616.0, colorless block, dimension 0.3×0.25×0.2 mm, 2196 reflections measured $(R_{int}=0.0234)$, 1978 unique, $wR_2=0.098$, conventional R=0.0401 on F values of 1485 reflections with $I>2\sigma(I)$, $(\Delta/$ σ)_{max}=000), S=1.02 for all data and 184 parameters. Unit cell determination and intensity data collection (2θ =50°) was performed on a Bruker P4 diffractometer at 293(2) K. Structure solutions by direct methods and refinements by full-matrix least-squares methods on F^2 . Programs: XSCANS (Siemens Analytical X-ray Instrument Inc.: Madison, WI, USA, 1996) for data collection and data processing; SHELXTL-NT (Bruker AXS Inc.: Madison, Wisconsin, USA, 1997) for structure determination, refinements, and molecular graphics. Further details of the crystal structure investigation can be obtained from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge, CB2 1EZ, UK (CCDC deposition no. of 9e: 609070).
- 9. Wang, S.; Tan, T.; Li, J.; Hu, H. Synlett 2005, 2658.
- Lee, M. J.; Kim, S. C.; Kim, J. N. Bull. Korean Chem. Soc. 2006, 27, 140.